

Enhanced structural and optical properties of ZnO nanorods synthesized via a modified sol-gel method

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A photo-anode semiconducting material with a nanorod geometry, hierarchical morphology, large surface area and high purity would yield good electron transport, effective dye loading and higher photon harvesting yielding a higher solar photon conversion [1]. The synthesis of these nanostructures could be achieved through sol-gel synthesis, hydrothermal synthesis, physical or chemical vapor deposition techniques, chemical bath deposition or electrochemical deposition etc. [2-5]. Among these methods, sol-gel method provides a facile and inexpensive synthesis of one-dimensional nanostructure without the use of sophisticated equipments or rigorous conditions [6].

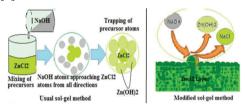


Figure 1: Usual and modified sol gel method

The oxidative etching modified sol-gel method reduces the random reaction of precursors, leading to a slow and controlled nucleation process.

We shall herein report a facile, environmental friendly, modified sol-gel synthesis route to hierarchical ZnO nanorods of high purity and large surface area, which could enhance the solar conversion efficiency of dye sensitized solar cells.

The as-synthesized ZnO nanorods were characterized using X-ray diffraction (XRD), Field emission scanning electron microscope (FESEM), UV-Vis spectroscopy (UV-Vis), Photoluminescence spectroscopy (PL) and BET analysis, which revealed impurity-free, highly crystalline, mesoporous and large surface-area, wurtzite phase ZnO nanorods with a diameters of

40-55 nm and a length of 200-225 nm, having a surface area of 152 m²g⁻¹. The characterizations have been investigated keeping the prime factors of efficient photo-anodes into considerations.

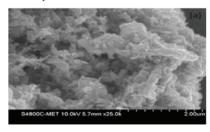


Figure 2: FESEM micrographs of ZnO nanorods

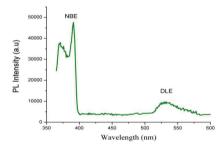


Figure 3: PL spectrum of ZnO nanorods

References

- 1. Hosokawa, M., Nogi, K., Naito, M., & Yokayama,
- T. Nanoparticle technology handbook. 2007.
- 2. Hung, C. H., & Whang, W. T. (2003). Materials Chemistry and Physics, 82(3), 705-710.
- 3. Dhafina, W. A., Ghapur, E. A., & Hasiah, S. (2012).ARPN Journal of Science and Technology, 2 (5), 432-436
- 4. Zhang, Q., Dandeneau, C. S., Zhou, X., & Cao, G. (2009). Adv. Mater. 21(41), 4087-4108.
- 5. Thavasi, V., Renugopalakrishnan, V., Jose, R., & Ramakrishna, S. (2009). R: Reports, 63(3), 81-99.
- Law, M., Greene, L. E., Johnson, J. C., Saykally, R.,
 Yang, P. (2005). Nature materials, 4(6), 455-459.