

TiO₂ – V₂O₅ nanocomposites for energy storage application

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Supercapacitor, the most promising energy storage system, can store substantially more energy than a conventional battery. According charge storage mechanism. supercapacitors can be classified into two categories. The first category is electrical double-layer capacitors (EDLCs) pseudocapacitors (or redox-capacitors). High cycle life (500,000 cycles) and good stability make supercapacitors more useful for the modern generation. Among various pseudocapacitor electrode materials, hydrous RuO₂ exhibits the most promising performance to date. However, the high cost of RuO₂ has limited its commercial attractiveness for supercapacitor applications. Therefore, there has been an extensive interest in developing alternative pseudocapacitor electrode materials, such as MnO₂, Co₃O₄, NiO , V₂O₅, TiO₂-V₂O₅ etc.[1].

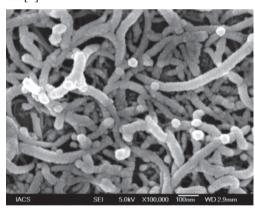
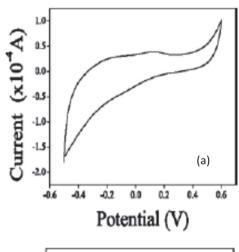


Figure 1: The FESEM images of TiO2-V2O5 films

Among all the metal-oxides, TiO₂-V₂O₅, particularly with proper molar ratios of Ti/V, has the advantage of moderate conductivity, low cost and can be easily synthesized. Herein, we have prepared the optimized TiO₂-V₂O₅ nanocomposites with different Ti/V molar ratios (10:10, 10:15 and 10:20). Moreover, the effect of the crystallinity and microstructure of TiO₂-V₂O₅ composites were examined with the help of XRD and FESEM followed by an insight into the cycling performance of TiO₂-

 V_2O_5 electrode for supercapacitor electrode materials. The Sp. Capacitance The cyclic voltammetry curve of TiO_2 - V_2O_5 composite with molar ratio of Ti:V=10:20 giving Sp. Capacitance value of 50 F/gm in 1M KCl electrolyte.



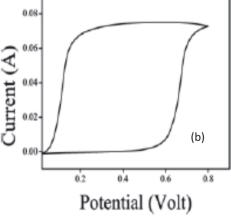


Figure 2: The CV images of TiO_2 - V_2O_5 films obtained from molar ratio Ti:V=10:20 using (a) 1MNaOH and (b) 1M KCl solution

Reference

1. C.D. Lokhande, D.P. Dubal, Oh-Shim Joo, CurrentApplied Physics 11 (2011) 255-270